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CHEMISTRY NOTES " Chapter 10 States of Matter Goals : To gain an understanding of : 1.. The kinetic theory of matter. 2. Avogadro's hypothesis.

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Modern Chemistry Chapter 6 Chemical Bonding . Chemical Bond "A link between atoms that results from the mutual attraction of their nuclei for electrons "Electrostatic attraction between proton and electron "Classified by the way the valence e- are distributed around nuclei of combined atoms .

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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