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Name _____ Worksheet: Periodic Trends Period _____ 1. Which statement best describes Group 2 elements as they are considered in order from top to bottom of the

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Title: Microsoft Word - 4-14-Periodic Table Trends Wkst.doc Author: Brent White Created Date: 7/6/2005 8:28:09 PM

Worksheet: Periodic Table Trends Name

Explanation: According to periodic trends, atomic radius increases from right to left on the periodic table. Therefore, we would expect Nitrogen to be larger than Oxygen. 3.

Periodic Trends - Killingly High School

Periodic(Trends(Worksheet(((Answer Key 1. Circle)the)element)withthe)largest)atomic)radius)andput)a)square)aroundthe)element)withthe)smallest)atomic)radius:))) Cu) K) Ni) Br))) Largest " K Smallest - Br) Explain why)you)made these choices.) Atomic radius decreases as you go left to right across a period. Potassium is in the far

Periodic Trends Worksheet Answers

The Periodic Table. A Periodic Table defined is a horizontal arrangement comprising of all the chemical elements that are present in the order of their atomic number, electronic configuration, and persisting chemical properties.

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Periodic Table Trends Worksheet Answers Chemistry

Fill in the Blank: Use the word bank provided to complete the sentences. nonmetal more family group left periodic table

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Directions: Use your Periodic table to complete the worksheet. 1. What is the atomic symbol for silver? 2. What is the atomic mass of mercury? 3. Ni is the symbol for what element? 4. The element that has the atomic number 17 is? 5. List the symbols for two transition metals. 6. Cu, Ag, and Au are all in what group #

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